Weekly Distribution of Syllabus

Class-11th Subject- Chemistry

Subject- Cnemistry					
Month	Week	Name of Chapter	Topics/Content		
	2-1/46 horse 40 h	Come Davis Comments of Chamilate	Gen. Introduction: Importance & Scope		
June	3rd (16 June - 19 June)	Some Basic Concepts of Chemistry	Historical Approach & Law of Chemical Combination		
			Dalton's Atomic Theory, Concept of Elements, atoms & Molecules		
	44h (24 have - 26 have)	Some Basic Concepts of Chemistry	Atomic & Molecular Masses, Mole Concept & Molar Mass		
	4th (21 June - 26 June)		Percentage Composition, Empirical & Molecular Formula, Stoichiometry		
		Structure of Atom	Discovery of Electron, Proton & Neutron		
	5th (28 June - 03 July)	Structure of Atom	Atomic Number, Isotops, Isobars & Thomson's Model of Atom		
			Rutherford's & Bohr's Model of Atom, Dual Nature of Matter De Broglie's Relationship, Heisenberg's Uncertainity Principal, Concept of Orbitals		
	1st (05 July - 10 July)		Quantam Numbers, Shape of Orbitals		
		Structure of Atom	Quantam Numbers, snape of Orbitals Rules for filling Electrons, Electronic Configuration		
		Classification of Elements & Periodicity	Significance of Classfication, History of Development		
		Classification of Elements & Periodicity	Present form of Periodic Table & Periodic Trends- Atomic Radii, Ionic Radii, Inert Gas Radii		
	2nd (12 July - 17 July)		Periodic Trends- I.E., E.G.A., Electronegetivity, Valency & Nomenclature Elements >100		
July		Chemical Bonding & Molecular Structure	Valence Electrons, Ionic Bond, Covalent Bond & Bond Parameters		
	3rd (19 July - 24 July)	Chemical Bonding & Molecular Structure	Lewis Structure. Polar Character of Covalent Bond & Covalent Character of Ionic Bond		
			V.B.T., Resonance & Geometry of Covalent Molecule		
1			VESPAR Theory, Concept of Hybridization & MOT		
	4th (26 July - 31 July)	Revision and SAT			
	4th (20 July - 31 July)				
	1st (02 Aug - 07 Aug)	States of Matter: Gases & Liquids	Three States of Matter, Types of Bonding, M.P. & B.P., Role of Gas Laws		
			Boyle's Law, Charle's Law, Gay Lussac's Law		
			Avagadro's Law, Ideal Behaviour, Derivation of Gas Equation, Avagadro No.		
	2nd (09 Aug - 14 Aug)	States of Matter: Gases & Liquids	Liquefication of Gases, Critical Temp., K.E. & Molecular Speed		
			Liquid State- V.P., Viscosity, Surface Tension		
		Chemical Thermodynamics	System & Surruondings, Work		
A	3rd (16 Aug - 21 Aug)	Chemical Thermodynamics	Heat, Energy, Extensive & Intensive Properties, State Function		
August			Ist Law Thermodynamics, Measurements of ΔU, ΔH & Hess Law		
			Enthalpy of Bond Dissociation, Combustion, Formation, Atomization, Sublimation		
	4th (23 Aug - 28 Aug)	Chemical Thermodynamics	Solution & Dilution, Entropy, Gibbs Free Energy Change		
			Criteria of Equilibrium, Second Law of Thermodynamics		
		Equilibrium	Equilibrium in Phy. & Chemical Process, Dynamic Nature of Equilibrium		
	5th (30 Aug - 04 Sept)	Equilibrium	Law of Mass Action, Equilibrium Contant, Factors affecting Equilibrium		
			Le Chatelier's Principle, Ionic Equilibrium		
			Degrree of Ionization, Ionization of Poly Basic Acids, Acid Strength		
	1st (06 Sept - 11 Sept)	Equilibrium	Concept of pH, Henderson Equation, Hydrolysis of Salts		
			Buffer Solutions, Solubility Product & Common Ion Effect		
		Redox Reactions	Concept of Oxidation & Reduction, Redox Reactions		
	2nd (13 Sept - 18 Sept)	Redox Reactions	Redox Reactions, Oxidation Number		
			Balancing Redox Reactions & Application of Redox Reactions		
Sept.		Hydrogen	Position of Hydrogen, , Occurance, Isotops		
sept.	3rd (20 Sept - 25Sept)	Hydrogen	Prepration, Properties & Uses of Hydrogen		
			Hydrides- Ionic, Covalent & Iterstitial, Water		
			Heavy Water & Hydrogen Peroxide		
	4th (27 Sept - 02 Oct)		Revision and SAT		

Weekly Distribution of Syllabus

Class-12th

Subject- Chemistry

Month	Week	Name of Chapter	Topics/Content
		Solid State	Classification of solids, Amorphous & Crystalline Solids
	3rd (16 June - 19 June)		Unit Cell in 2D & 3D lattice, Density Calculation
	, , ,	F	Packing, P.E., Voids, No. of Atoms per Unit Cell
			Defects, Electrical & Magnetic Properties, Band Theory
June	4th (21 June - 26 June)	Solid State	Conductors, Semiconductors, Insulators & n-p type Semiconductors
	, , ,	Solution	Types of Solutions, Concentration & Solubility of Gases
		Solution	Colligative Properties - RLVP & Raoult's Law
	5th (28 June - 03 July)		Elevation in B.P., Depression in F.P. & Osmotic Pressure
			Determination of Mol. Mass , Abnormal Mol. Mass, Van't Hoff Factor
		Electrochemistry	Electrochemical Cell, Electrode Potential & E.M.F. of Cell
	1st (05 July - 10 July)		Electrochemical Series, Conductivity
			Kohlrausch's Law & Electrolysis
		Electrochemistry	Betteries & Corrosion
	2nd (12 July - 17 July)	Chemical Kinetics	Rate of Rxn., Factors affecting Rate of Rxn. & Rate Constant
July			Order and Molecularity of Rxn.& Speccific Rate Constant
			Integrated Rate Equation & Half Life
	3rd (19 July - 24 July)	Chemical Kinetics	Concept of Collision Theory, Activation Energy & Arrhenious Equation
		Surface Chemistry	Adsorption, Factor affecting Adsorption & Catalysis
	4th (26 July - 31 July)	Revision and SAT	
	1-+ (02 Aug 07 Aug)	Surface Chemistry	Enzyme Catalysis & Colloidal State
	1st (02 Aug - 07 Aug)		Purification & Properties of of Colloids
			Origin of Charge on Colloidal Particals & Emulsions
		General Isolation & Process	Metallurgy & Refining of Metals
	2nd (09 Aug - 14 Aug)		Chromatographic Methods & Ellingham Diagram
			Extraction - Fe, Cu & Ag
. .		General Isolation & Process	Extraction - Au, Al & Zn
August	3rd (16 Aug - 21 Aug)	p-Block Elements	Group - 15 (General introduction, Physical & Chemical Properties)
			Group - 15 (Nitrogen & Phosphorus)
		p-Block Elements	Group - 16 (General introduction, Physical & Chemical Properties)
	4th (23 Aug - 28 Aug)		Group - 16 (Oxygen & Sulphur)
			Group - 17 (General introduction, Physical & Chemical Properties)
		p-Block Elements	Group - 17 (Chlorine & Interhalogen Compound)
	5th (30 Aug - 04 Sept)	p block Elements	Group - 18 (Noble Gases)
		d & f Block Elements	General Introduction, Charactristics of Transition Metals
			General Trends in Properties of Transition Metals
	1st (06 Sept - 11 Sept)	d & f Block Elements	Prepration & Properties of K2Cr2O7 and KMnO4
		[Lanthanoids - General & Lanthanoid Contraction
		d & f Block Elements	Actinoids- General & Comperision with Lanthanoids
	2nd (13 Sept - 18 Sept)	Coordination Compounds	Coordination Compounds- General, Ligands & Magnetic Properties
C			Coordination Compounds - IUPAC Nomenclature
Sept.			Bonding, Werner's Theory, VBT & CFT
	3rd (20 Sept - 25Sept)	Coordination Compounds	Colours in Coordination Compounds & Stability of Complex
		· · · · · · · · · · · · · · · · · · ·	Metal Carbonyls & Applications of Coordination Compounds
	4th (27 Sept - 02 Oct)	Revision and SAT	